

riag Oberflächentechnik AG · Postfach 169 · CH-9545 Wängi TG

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# riag Sn 865 HS

#### Methansulfonate-based satin tin process

riag Sn 865 HS is a low foaming high speed process that can be used for decorative as well as technical applications also for belt conveyer systems. It is characterised by an extremely homogeneous layer thickness distribution, regular appearance, minimised whisker growth properties and exceptionally low sludge formation in the electrolyte.

# **Properties**

- satin mat deposits across a wide current density range
- excellent coat thickness distribution
- excellent solderability of coats
- technical and decorative application

#### Make up

riag Sn 860 Tin riag Sn 860 Acid riag Sn 865 Tenside riag Sn 865 Additive riag Sn 865 Antiox

Range				Optimum	
200	_	265	mL/L	220	mL/L
70	_	100	mL/L	70	mL/L
40	_	80	mL/L	50	mL/L
3.5	_	7	mL/L	5	mL/L
1	_	2	g/L	1.5	g/L

#### Make up

The tank is filled with deionised water to 50 % of the final volume. While stirring you add the riag Sn 860 Acid and the riag Sn 860 Tin. Then add the required amount of riag Sn 865 Additive, riag Sn 865 Tenside and riag Sn 865 Antiox (predissolved). The electrolyte is filled up with water to the final volume. First some dummy parts are coated.

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# **Operating values**

 Range
 Optimum

 Tin (II)
 60 - 80 g/L
 66 g/L

 riag Sn 860 Acid
 70 - 100 mL/L
 80 mL/L

### **Operating parameters**

Temperature:  $45 \,^{\circ}\text{C} \, (40 - 60 \,^{\circ}\text{C})$ 

Cathodic current density:  $5 - 40 \text{ A/dm}^2$ 

Ratio anodes to cathodes: minimum 1:1

Current efficiency: < 100 %

Deposition rate: at 10 A/dm<sup>2</sup> approx. 5 μm/min.

Anodes: The purity of the tin anodes should at least be 99.99 %. We recommend

the use of polypropylene anode bags.

Tanks: Plastic or lined steel

Filtration: For high performance electrolytes constant filtration is necessary.

5 µm polypropylene filter cartridges are generally sufficient.

Heating: Thermostatic controlled temperature regulation is essential

Fume extraction: Recommended

Preparation of New tanks should be treated with **riag Sn 860 Acid** and

new tanks: riag Sn 865 Tenside for 24 hours.

When a conversion of the tank from a lead containing electrolyte takes place, an alkaline primary cleaning is recommended. Our sales staff will

gladly advise you.

Maintenance: Analyse and adjust riag Sn 860 Tin and riag Sn 860 Acid as well as

riag Sn 865 Antiox regularly. The riag Sn 860 Tin solution contains 300 g/L tin and 50 mL/L riag Sn 860 Acid. To increase the tin content of the electrolyte by 1 g/L 3.3 mL/L riag Sn 860 Tin solution are required.

Dosage of riag Sn 865 Additive and riag Sn 865 Tenside are

determined by ampère hours.

Usage: The additives are used up by drag out as well as electrochemical, that is

by anodic or cathodic processes. Therefore the usage may vary process-

related.

**riag Sn 865 Tenside** 2.5 – 5.0 L/10 kAh

**riag Sn 865 Additive** 3.0 – 5.0 L/10 kAh

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General:

In particular drag-in of chloride into the tin electrolyte has to be avoided. Therefore the parts are activated with **riag Sn 860 Acid** (approx. 5 % V) instead of hydrochloric acid.

Brass and other zinc containing alloys must not at all tin-plated directly since zinc diffuses into the tin coat. In this case a barrier coat of copper or nickel is required. **riag Sn 865 Antiox** prevents the formation of Sn (IV) and the following clouding of the electrolyte.

# **Environmental considerations and product safety**

All concentrates, rinse waters and waste solution must be treated and discharged in accordance with local effluent control regulations. Information can be gleaned from the material safety data sheets. Chemicals shall not be stored below 10 °C.

## Liability

This instruction manual was compiled with reference to the state of the art and all current standards, and is based on the long-term knowledge and experience of riag. However, riag cannot monitor compliance with this instruction manual and the methods described herein at the customer/end-user's premises. Work carried out with riag products must be adapted accordingly to meet local conditions. In particular, riag cannot accept liability for damage, loss or cost incurred due to a failure to adhere to this instruction manual, improper application of the methods, unauthorised technical modifications, insufficient maintenance or the absence of maintenance in respect of the requisite technical hardware or equipment, or in the event of use by unqualified personnel. riag is not liable for damage or loss caused by riag or its employees except where intention or gross negligence can be proved. riag furthermore reserves the right to make changes in relation to products, methods and the instruction manual without prior notice.

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# **Analysis (Analytical Methods)**

Sample preparation:

The sample must be taken from a well-mixed point.

Tin (II)

Reagents: lodine 0.05 mol/L

Hydrochloric acid 37 % Starch solution 1 % Calcium carbonate p.a.

Procedure: 5 mL electrolyte are transferred via pipette into a

250 mL beaker, add

50 mL deion. water, add

40 mL hydrochloric acid 37 %, add

approx. 2 g calcium carbonate, add

approx. 2 mL starch solution

Titrate with iodine 0.05 moL/L from colourless to dark

blue. The dark blue colour has to stay for 30 s

Calculation: Use in mL x 1.186 = g/L Tin(II)

riag Sn 860 Acid

Reagents: Sodium hydroxide solution 1 mol/L

Methyl red 0.2 % in ethanol

Procedure: 5 mL electrolyte are transferred via pipette into a

100 mL beaker, add

ca. 50 mL deion. water

ca. 3 drops methyl red

Titrate with sodium hydroxide from orange-red to yellow

Calculation: Use in mL x 20.3 = mL/L riag Sn 860 Acid

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